

Consultancy – Paper Industry



Stability Improvements

In

Clay Slurry Prepⁿ Process

Clay Slurry Prepⁿ – pH control

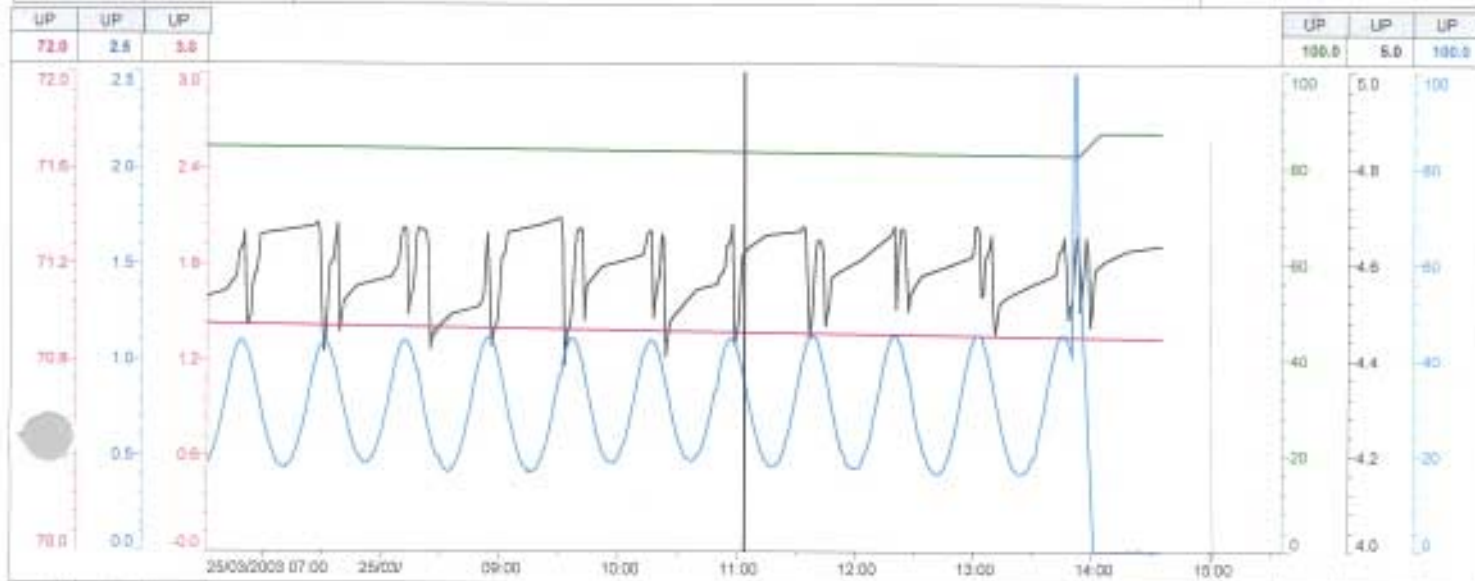


4 COMMON:3 TRENDS:trendpt6 - Caldonian

CPMQMS01: 25/03/2003 14:35:00

Tag Name	EDU	Description	Ruler	Delay	Filter	Current	Mark	Symbol
134F826	L/S	FILLER FLOW TO STATIC MIXER	5.591643E-03	0	0	5.242382E-03		□
134L828	PCT	LEVEL OF FILLER REACTION TANK	82.83049	0	0	87.18623		○
134FC827	L/S	WATER FLOW TO FILLER STATIC MIXER	2.976378E-02	0	0	2.189868E-03		▽
134QC846	PH	PH OF FILLER CLAY	4.598499	0	0	4.638361		△
134LC834	PCT	LEVEL OF FILLER SCREENING TANK	70.90332	0	0	70.8896	✓	■
134LC834-op	PCT	LEVEL OF FILLER SCREENING TANK	22.18979	0	0	0		●

Time Base	No of Points	Selection Criteria	Ruler Time
1MIN	1000		25/03/2003 12:48:53

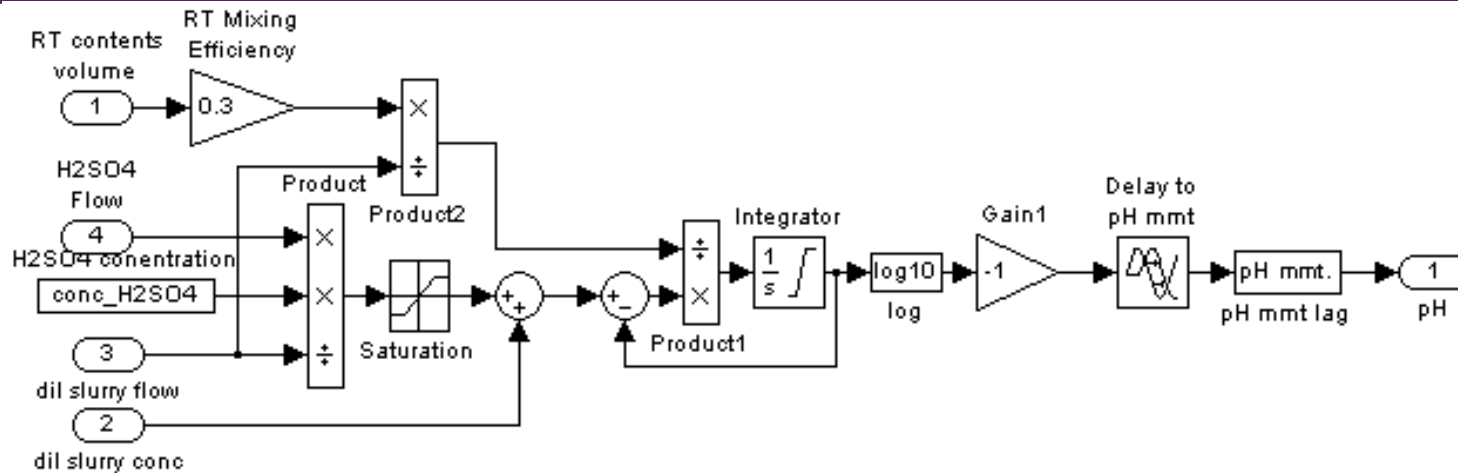


76.0	0.0	0.0
Down	Down	Down
25/03/2003 08:32:00		

<<	<	Zoom In	Zoom Out	>	>>
Statistics...	Time Base In	Time Base Out	Current Time	Red Tag	

0.0	-4.0	0.0
Down	Down	Down
25/03/2003 15:36:00		

Titration Model (pH calculation)



$$Q_{RT} * C_{RT} = Q_{H_2SO_4} * C_{H_2SO_4} + Q_{DS} * C_{DS} \quad (\text{neglecting RT mixing lag effect})$$

$$C_{RT} = \frac{Q_{H_2SO_4} * C_{H_2SO_4} + Q_{DS} * C_{DS}}{Q_{RT}}$$

but $Q_{RT} \approx Q_{DS}$

$$\therefore C_{RT} = \frac{Q_{H_2SO_4} * C_{H_2SO_4}}{Q_{DS}} + C_{DS}$$

or $C_{RT} = \frac{1}{(1 + \tau S)} * \left(\frac{Q_{H_2SO_4} * C_{H_2SO_4}}{Q_{DS}} + C_{DS} \right)$ (with RT mixing lag effect)

where $\tau = \frac{\text{Volume of RT contents (t)} * \text{RT Mixing Efficiency}}{Q_{DS}(t)}$

$$pH = -\log_{10} \left[\frac{1}{(1 + \tau S)} * \left(\frac{Q_{H_2SO_4} * C_{H_2SO_4}}{Q_{DS}} + C_{DS} \right) \right]$$

Where :

RT = Reaction Tank

Q_{RT} = RT Effluent Volumetric Flow (m^3/s)

C_{RT} = Ion Concentration of RT Effluent Flow ($\#/m^3$)

$Q_{H_2SO_4}$ = H_2SO_4 Volumetric Flow (m^3/s)

$C_{H_2SO_4}$ = Ion Concentration of H_2SO_4 ($\#/m^3$)

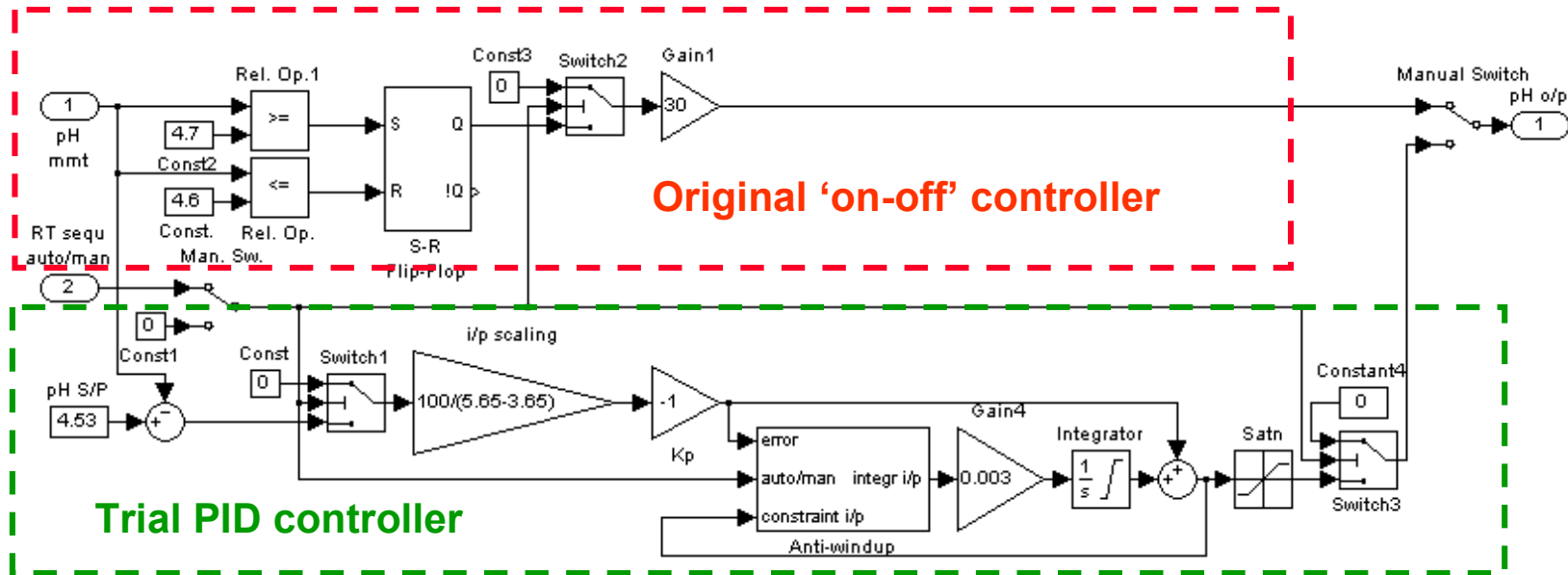
Q_{DS} = Dilute Slurry Volumetric Flow (m^3/s)

C_{DS} = Ion Concentration of Dilute Slurry ($\#/m^3$)

τ = First Order RT Mixing Lag

pH = RT Effluent pH

pH Control Evaluation



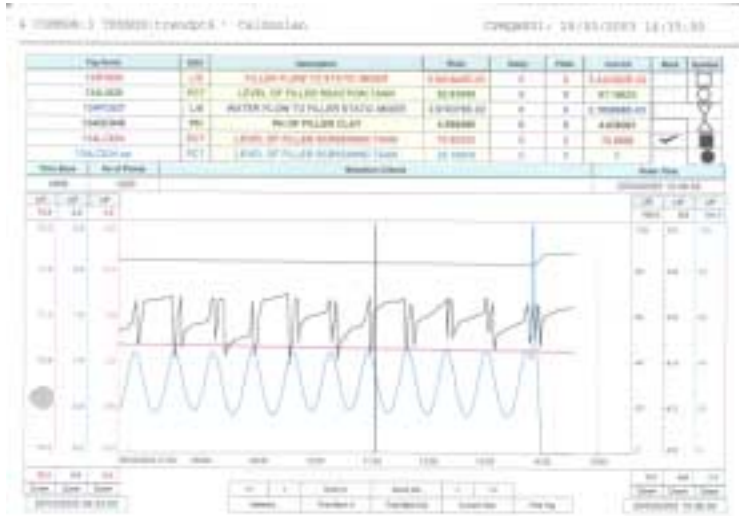
Modelled Controller Features

- Ideal PID Controller Structure
- Logic to inhibit controller during RT empty cycle
- Logic to switch between PID & 'bang-bang' controllers
- Current (dilution water) valve requires ~2% open operating point! (for PID controller)

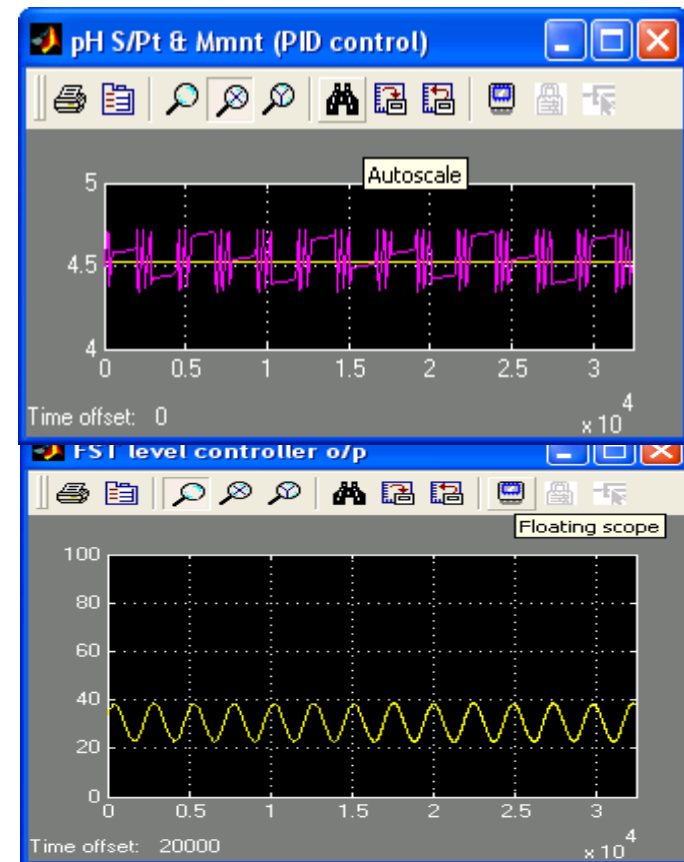
Validation of Model pH Dynamic Data



Plant data over a 9 hour period



Model data over a 9 hour period

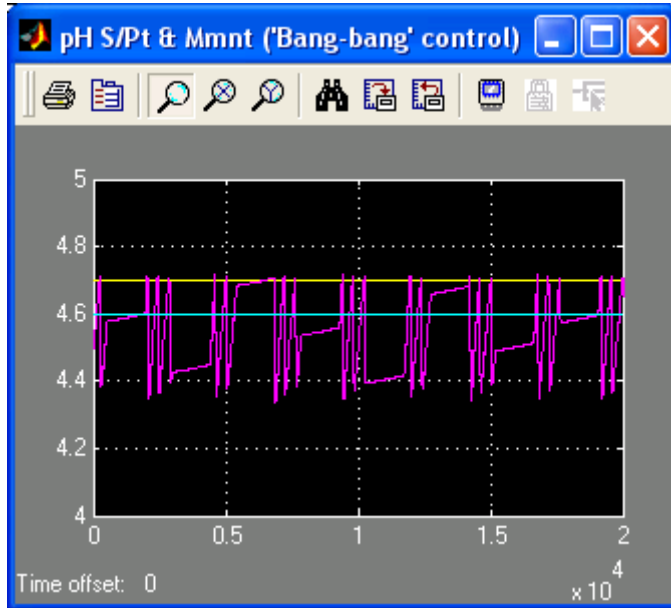


Imperatives for PID pH Control

- Key Issues:

- H₂SO₄ flow must be stopped during RT empty cycle
 - logic from RT level limit switches to close/open valve
- pH control must be suspended during RT empty cycle
 - the integral action must be suspended (for empty cycle)
- (Current) oversized H₂SO₄ valve must be changed
 - select valve in (max. C_v) range of 3 to 6
- Start-up PID parameters (from model) must be used

pH Control Comparison

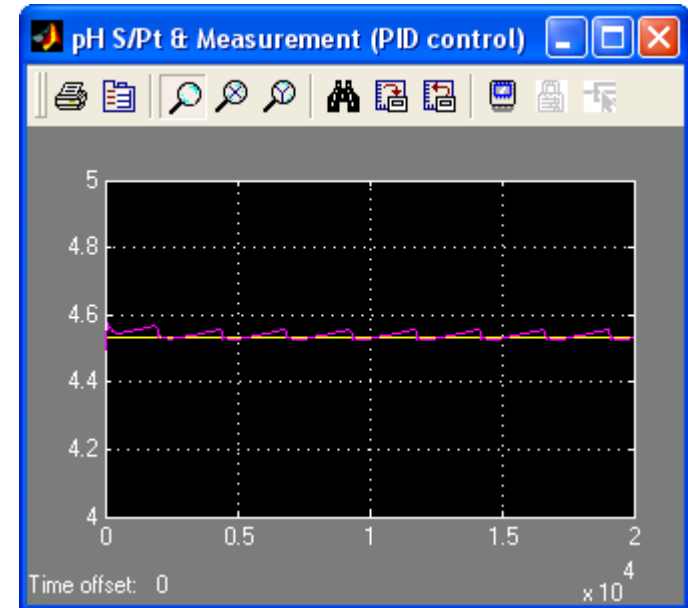


Current 'On-Off' Control

pH 'Trigger' points at

- 4.7 (H_2SO_4 on)
- 4.6 (H_2SO_4 off)

Peak to peak pH amplitude
0.38 pH units



PID Control Alternative

pH set point at
- 4.53

Peak to peak pH amplitude
0.03 pH units

Reduced by >90%

